

# **37. Analytical techniques**

## **37.2 Gas - liquid chromatography**

### **Paper 4**

Marking Scheme

**Q1.**

(b)(i)	time between injection and detection	<b>1</b>
(b)(ii)	29.5%	<b>1</b>

**Q2.**

(c)(i)	time between injection and detection [1]	<b>1</b>
(c)(ii)	an unreactive gas <b>AND</b> a non-polar liquid [1]	<b>1</b>
(c)(iii)	area of peak divided by total area of all peaks $\times 100\%$ [1]	<b>1</b>

**Q3.**

(g)	mass of L = $58 / 44 \times 5.52 \times 10^{-2} = 7.28 \times 10^{-2} \text{ g}$ conc. of L = $7.28 \times 10^{-2} / 116 = 6.27 \times 10^{-4} \text{ (mol dm}^{-3}\text{) min 2sf}$	<b>1</b>
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**Q4.**

(a)(i)	<b>stationary:</b> non-volatile (non-polar) liquid <b>mobile:</b> nitrogen / argon / any inert gas	<b>1</b>
(a)(ii)	the time that the substance stays in the column <b>OR</b> time between injection and detection <b>OWTTE</b>	<b>1</b>
(a)(iii)	areas are 30, 100, 25 <b>OR</b> 30, 150, 25 ( $\frac{1}{2} \times b \times h$ ) <b>OR</b> 100 / 155 [1] % of B = $100 \times (100 / 155)$ <b>OR</b> % of B = $100 \times 150 / 205 =$ = <b>64.5</b> = <b>73.2</b> [1] ecf M1 <i>areas</i>	<b>2</b>